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Journal of Macromolecular Science, Part A

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713597274

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To cite this Article Tang, A. C. , Li, Z. S. and Sun, C. C.(1992) 'On The Intramolecular Cyclization of A_a Type Crosslinking Reaction', Journal of Macromolecular Science, Part A, 29: 11, 277 – 282 **To link to this Article: DOI:** 10.1080/10101329208054592 **URL:** http://dx.doi.org/10.1080/10101329208054592

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ON THE INTRAMOLECULAR CYCLIZATION OF A TYPE CROSSLINKING REACTION A. C. Tang, Z. S. Li, and C. C. Sun Institute of Theoretical Chemistry Jilin University, Changchun, Jilin, China

ABSTRACT

By taking into consideration of intramolecular cyclization, the A type crosslinking reaction is investigated to give the sol fraction for postgelation from which the equilibrium number fraction distribution is deduced. Furthermore, by means of differentiation technique, a recursion formula for evaluating the kth polymer moment is obtained.

INTRODUCTION

As is well known, the crosslinking theory for polymers was initiated by Flory-Stockmayer[1-3]. For a system involving intramolecular cyclization, the reaction was investigated by Harris, Kilb, Gordon, Ahmad, and Stepto[4-8]. In this paper, an alternative way is proposed to approach the A₂ type crosslinking reaction in which the intramolecular cyclization is involved. The sol fraction for a post-gel is studied in detail to deduce the eqilibrium number fraction distribution with Flory-Stockmayer distribution as a criterion. In the theory of branching processes, a probability generating function method has been proposed by Gordon[9,10] for the evaluation of polymer moments. In this paper, by means of the number distribution characterized by intramolecular cyclization, a direct differentiation technique is used to obtain a recursion formula that holds true for both pregelation and postgelation states in evaluating the kth polymer moment explicitly.

1. SOL FRACTION FOR POSTGELATION

Let us consider a crosslinking system associated with monomer A having a-functionality. Let p and p' be the total and sol equilibrium fraction conversions, respectively. Furthermore, each of the conversions can be separated into two parts in the forms

$$p = q + f$$
 (1)
 $p' = q' + f'$ (2)

where

q = fraction of functional groups that have reacted intermolecularly

- q'= fraction of functional groups that have reacted intermolecularly in the sol
- f = fraction of functional groups that have reacted intramolecularly
- f' = fraction of functional groups that have reacted intramolecularly in the sol.

Note that the relation in Eq.(1) was proposed by Gordon[7] in treating ring-chain competition kinetics in linear polymers.

From probability consideration, each of the functional group of monomer A_{a} may have three kinds of probabilities in the sol for postgelation such that

- 1. The probability of finding a functional group in the system that is not involved in crosslinking is 1-q-f.
- The probability of finding a functional group in the system which joins crosslinks with a certain n-mer in the sol is qS(1-f'-q')/(1-f-q), where S represents the sol fraction which varies from 1 to 0 for postgelation.
- 3. The probability of finding a functional group in the system which participates in intramolecular cyclization in the sol is f'S^{1/a}.

Since the sol fraction S can be interpreted as the probability of finding a monomer in the sol, it is not difficult to find that the sol fraction S can be expressed in terms of the three kinds of probabilities mentioned above in the form

$$S = \left(1 - f - q + qS\frac{1 - f' - q'}{1 - f - q} + f'S^{1/a}\right)^{a}$$
(3)

For brevity, we only state without argument that from probability consideration, an alternative expression of S can be obtained

$$S = \left(\frac{1 - f - q}{1 - f' - q'}\right)^{a}$$
(4)

It should be noted that the two expressions of S given by Eqs.(3) and (4) are independent.

In the above discussion, we have introduced 7 quantities (q,q',f, f',p,p',S) subject to the 4 relations given in Eqs.(1), (2), (3) and (4), and thus only three of the 7 quantities are independent. When the three independent quantities taken as observables are suitably chosen, such as p, p' and S, the remaining 4 can be evaluated without difficulty.

Now let us discuss the gel point of A type crosslinking reaction involving intramolecular cyclization.^aApplying Eq.(4) to the term (1-f'-q')/(1-f-q) appearing in the right hand side of Eq.(3) gives

. . . .

$$S = (1 - f - q + qS^{(a-1)/a} + f'S^{1/a})^{a}.$$
 (5)

By differentiating the both sides of Eq.(5) with respect to S, and then, by considering the critical condition $S \rightarrow 1$ with $f=f'=f_c$, we obtain

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$$q_c = (1 - f_c)/(a - 1)$$
 (6)

where the symbol c signifies the critical point of the sol-gel transition. Thus, the gel point p_{i} involving intramolecular cyclization can be easily written as, by means of Eqs.(1) and (6)

$$p_{c} = f_{c} + q_{c} = f_{c} + (1 - f_{c})/(a - 1) .$$
(7)

If the intramolecular cyclization conversion in the sol is not considered, the gel point p_c in Eq.(7), by putting $f_c=0$, can be reduced to the form

$$p_{c}(f_{c}=0) = 1/(a-1)$$
 (8)

This form is the well known gel point without cyclization obtained by Flory-Stockmayer[1,2]. It is clear that the inequality

$$p_{c} > p_{c}(f_{c}=0)$$
(9)

holds good. This inequality agrees with the experimental fact that the observed p is larger than 1/(a-1). When a=2, both gel points p and p (f=0) given by Eqs.(7) and (8) take the same value of unity. This means that no gelation occurs for the case of a=2, no matter whether intramolecular cyclization is considered or not.

2. EQUILIBRIUM NUMBER FRACTION DISTRIBUTION Let us rewrite the sol fraction S in Eq.(4) in the form

$$S = F(z) = (1 - f - q)^{a} z^{a}$$
 (10)

with

$$z = 1/(1 - f' - q')$$
 (11)

By combining the two independent expressions of sol fraction given by Eqs.(3) and (4), an alternative expression of z can be obtained

$$z = \frac{1}{1 - f'} + t \phi(z)$$
 (12)

with

$$t = \frac{q(1 - f - q)^{a-1}}{(1 - f')(1 - f - q)}, \quad \phi(z) = z^{a-1}. \quad (13)$$

By means of Lagrange's expansion theorem[11], the sol fraction S=F(z) in Eq.(10) can be expanded as a power series in t by the formula

$$S=F(z)=F(\frac{1}{1-f'}) + \sum_{n=2} \frac{t^{n-1}}{(n-1)!} \left\{ \frac{d^{n-2}}{dz^{n-2}} \left[\frac{dF(z)}{dz} (\phi(z))^{n-1} \right] \right\}_{z=\frac{1}{1-f'}}$$
(14)

This expansion formula is subject to the restriction condition

given by Eq.(12). A straightforward calculation leads us to the result, from Eq.(14)

$$S = \sum_{n=1}^{n} n^{p}$$
(15)

where \mathbf{P}_n , which is the equilibrium number fraction distribution of n-mer, takes the form

$$P_{n} = \frac{a(an-n)!}{n!(an-2n+2)!} \left(\frac{q}{1-f'}\right)^{n-1} \left(\frac{1-f-q}{1-f'}\right)^{an-2n+2}.$$
 (16)

Under the assumption that no intramolecular cyclization occurs in the sol(f'=0), the distribution P_n can be reduced, for pregelation, to

$$P_{n} = \frac{a(an-n)!}{n!(an-2n+2)!} q^{n-1}(1-q)^{an-2n+2}.$$
 (17)

This form is the well known Flory-Stockmayer equilibrium number fraction distribution. In obtaining the distribution in Eq.(17), we have made use of f=f'=0. For the case of postgelation, the distribution P in Eq.(16) can be reduced approximately to the one given by Eq.(17). For brevity, the detail is omitted here.

3. RECURSION FORMULA FOR THE KTH MOMENT The kth moment $\mathbf{M}_{\mathbf{k}}$ is defined as

$$M_{k} = \sum_{n=1}^{n} {}^{k}P_{n}$$
, $k = 0, 1, 2, ...$ (18)

where P_n is given by Eq.(16), in which the intramolecular cyclization is involved. By differentiating the both sides of Eq.(18) with respect to q, a recursion formula of polymer moments involving M_k and M_{k+1} can be deduced

$$M_{k+1} = \frac{1}{1 - f - (a-1)q} [q(1 - f - q) - \frac{\partial M_k}{\partial q} + (1 + q - f) M_k], \ k=0,1,\dots (19)$$

This recursion formula is suitable for both pre-gel and post-gel in evaluating the polymer moments explicitly.

From the probability meaning of the distribution P_n , it is not difficult to find that the zeroth moment M_o is the number of n-mer with n=1, 2, 3,... in the sol per total number of monomers in the crosslinking system, i.e.

$$M_{\circ} = \sum_{n=1}^{P} P_{n} = \begin{cases} 1 - \frac{aq}{2(1 - f)}, & \text{for pre-gel} \\ \\ S[1 - \frac{aq'}{2(1 - f')}], & \text{for post-gel} \end{cases}$$
(20)

From this expression, we can easily obtain, via the recursion formula (19), the first moment ${\rm M}_1$

$$M_{1} = \sum_{n=1}^{n} nP_{n} = \begin{cases} 1, & \text{for pre-gel} \\ s, & \text{for post-gel} \end{cases}$$
(21)

This expression is in agreement with that given by Eq.(15) which has been obtained by means of Lagrange expansion method. Taking M₁ as starting point, we can obtain M2, with consecutive recursion

$$M_{2} = \sum_{n=1}^{n} n^{2} P_{n} = \begin{cases} \frac{V_{2}}{1 - f - (a - 1)q}, & \text{for pre-gel} \\ \\ \frac{T_{2}}{f + (a - 1)q - 1}, & \text{for post-gel} \end{cases}$$
(22)

with

$$V_2 = 1 + q - f$$
 (23)

$$T_{2} = - [q(1 - q - f)\frac{\partial S}{\partial q} + (1 + q - f)S]$$
(24)

$$\frac{\partial S}{\partial q} = \frac{a(1 - S^{(a-1)/a})S^{(a-1)/a}}{(a - 1)qS^{(a-2)/a} - (1 - f')} .$$
(25)

From the expression of M_{2} which is taken as starting point, one can approach, by repeated application of the recursion formula (19), a general expression of the kth moment $M_{\rm k}$ for $k \ge 2$

$$M_{k} = \sum_{n=1}^{k} n^{k} p_{n} = \begin{cases} \frac{V_{k}}{[1 - f - (a - 1)q]^{2k-3}}, & \text{for pre-gel} \\ \frac{T_{k}}{[f + (a - 1)q - 1]^{2k-3}}, & \text{for post-gel} \end{cases}$$
(26)

where V_k and T_k are subject to the same recursion formula

$$H_{k} = (2k-5)(a-1)q(1-q-f)H_{k-1} + (1-f-(a-1)q)[q(1-f-q)-\frac{\partial H_{k-1}}{\partial q} + (1+q-f)H_{k-1}].$$
(27)

Note that H_k is used to denote either V_k for pre-gel or T_k for post-gel.

REFERENCES

- [1] Flory, P. J., Am. Chem. Soc., 63, 3083, 3091, 3096(1941).
- [2] Stockmayer, W. H., J. Chem. Phys., <u>11</u>, 45(1943).
- [3] Stockmayer, W. H., J. Polym. Sci., 9, 69(1952); <u>11</u>, 424(1953).
 [4] Harris, F. E., J. Chem. Phys., 23, 1518(1955).
- [5] Kilb, R. W., J. Phys. Chem., 62, 969(1958).
- [6] Gordon, M. and Scantlebury, G. R., J. Polym. Sci., Part C, 16, 3933(1968).
- [7] Gordon, M. and Temple, W. B., Makromol. Chem., 152, 277(1972).

- [8] Ahmad, Z. and Stepto, R. F. T., Colloid Polym. Sci., <u>258</u>, 663(1980).
- [9] Gordon, M., Proc. R. Soc., London, A268, 240(1962).
- [10] Dobson, G. R. and Gordon, M., J. Chem. Phys., 43, 705(1965).
- [11] Whittaker, E. T. and Watson, G. N., A Course of Modern Analysis, Reprinted, Cambridge University Press, Cambridge, 1952.